MICROSCOPY AND SPECTROSCOPY TOOLS FOR THE DESCRIPTION OF DELIGNIFICATION

H. M. HERNÁNDEZ-HERNÁNDEZ,*,***, J. J. CHANONA-PÉREZ,*** E. TERRÉS,**** A. VEGA,*****
P. LIGERO,***** R. R. FARRERA-REBOLLO*** and S. VILLANUEVA*

*CONACYT – Center for Research and Assistance in Technology and Design of the State of Jalisco A.C. (CIATEJ), Guadalajara 44270, Jalisco, Mexico

**Center for Research and Development in Food Agrobiotechnology, C.P. 42162, San Agustín Tlaxiaca, Hidalgo, Mexico

***Department of Biochemistry Engineering, National School of Biological Sciences, National Polytechnic Institute, Col. Santo Tomas C.P. 11340, Cd. de Mexico, Mexico

*****Mexican Petroleum Institute (IMP), Colonia San Bartolo Atepehuacan, C.P. 07730, Cd. de Mexico, Mexico

******Advanced Scientific Research Center (CICA), Department of Physical Chemistry and Chemical Engineering, Coruña University, 15071 A Coruña, Spain

™ Corresponding author: H. M. Hernández-Hernández, hhernandez@ciatej.mx

Received June 6, 2018

This work describes and evaluates the changes in agave fibres occurring during a green pulping process by microscopy and spectroscopy techniques, providing a comprehensive and quantitative view of the delignification of agave (*Agave atrovirens*) fibres. The effects of 80% acetic acid concentration and 0.3% hydrochloric acid concentration on pulp yield, kappa number and pulp viscosity were investigated over time. In order to study the influence of the Acetosolv treatment on the microstructure and chemical composition of agave fibres, spectroscopic and microstructural characterisations of selected pulps were performed at different points during the process.

Keywords: lignocellulosic material, confocal laser scanning microscopy, delignification, scanning electron microscopy, Raman spectroscopy

INTRODUCTION

Much agroindustrial waste comprises lignocellulosic materials, which are sources of important biomacromolecules, such hemicelluloses, lignin and cellulose. The usage of such renewable materials opens a new sector in the chemical market, with potential positive impacts on mankind and the environment.^{2,3} Currently, many works focus on the usage of these materials in various applications, such as power, composite precursors, sources of bioactive compounds, new applications for cellulose, biodegradable plastics and new uses for lignin.⁴⁻⁶ Cellulose, the most abundant renewable natural polymer found in plants, is synthesised as structural microfibrils in the cell walls with different arrangements.⁷ Cellulose fibres are found in a wide variety of raw materials, including wood, leaves, seeds, fruits, bark and

straws. Non-wood fibre sources are attractive due to their lower lignin content and their availability, given their shorter growing cycle.8 They have been widely used in a number of applications due to their favourable characteristics, such as renewability, biodegradability, high availability and sustainability. 6,9-12 Agave is grown in the semi-desert areas of Mexico, mainly in Central Valley, Hidalgo, Puebla and the Tlaxcala states. 13 Agave has been used for thousands of years in Mexico, with diverse applications as for the feeding, medication, clothing and construction.⁶ The lignocellulosic biomass of agricultural residues contains mainly cellulose (38%), hemicellulose (32%) and lignin (17%) as major components, which can be converted into useful bioproducts, adopting the concept of 'wealth from waste'.14

Agroindustrial lignocellulosic by-products are the most promising sources of cellulose and lignin. These include sugarcane bagasse, corn straw, palm-pressed mesocarp fibre, sisal, pineapple leaves *etc.* ^{15–19} Agave waste may be an alternative source of cellulose and lignin. ^{20,21} The leaves are cut to obtain the "*piña*" (pinecore) used for the making alcoholic beverages, like mezcal and tequila. ^{6,22} Repeatedly harvested leaves or discarded reused leaves offer a potentially continuous supply of feedstock of lignocellulosic biomass. ^{6,22}

Many processes and techniques have been developed in order to fractionate lignocellulosic biomass, among them, the Organosolv process uses organic acids. The list of possible green organic solvents used to fractionate biomass is extensive, for example, acetic acid, formic acid or ethanol, which are catalysed with strong acids (hydrochloric acid or sulfuric). 1-3,23,24 Organosolv process can 'clean' the pulp by two simultaneous mechanisms: delignification and solubilisation of the hemicelluloses;² this process has been used for the treatment of lignocellulosic materials. 1,3,25–27 Specifically, the Acetosolv process is used to fractionate biomass, taking advantage of the nature of acetic acid catalysed with hydrochloric acid. 23,25

With minimal environmental impact, the Acetosolv process can fractionate the lignocellulosic material into cellulose (pulp fibres), acid lignin and monosaccharides. The acid delignification process breaks the α -aryl ether bonds by means of hydrolysis.^{25,27}

Agave waste may be an alternative source of cellulose and lignin with several applications in food, but the delignification mechanisms at the microstructural level have been scarcely studied on the fibre and the interest in agave fibres has been increasing. Thus, the physical and chemical mechanisms that occur during the Acetosolv treatment of agave fibres and the relationship between the chemical composition and the microstructure of fibres are evaluated by microscopy and spectroscopy.

EXPERIMENTAL

Raw material and fibre isolation

In order to obtain the fibre, leaves of agave (*Agave atrovirens*) waste were collected from a 5-7 years old plantation established in Mexico City, Mexico. Slices of agave leaves (20 x 20 x 0.3 cm) were dried in a convective dryer at 60 °C and an air flow rate of 3 m/s.³² Subsequently, the dry slices were milled (Blade

mill, Moulinex, Selongey, France) and the fibres were separated through a mesh sieve no. 8 (US Standards). Fibres with a length of 2.36 mm or longer were stored in hermetic plastic bags at room temperature (20-22 °C) until analysis; the final moisture content was 8.6%.

Raw material characterisation

Residual lignin (T222 om88), moisture content (T412 om-11), solvent extractives (T204 cm-97), water solubility (T207 cm-99), ash (T211 om-02) and holocellulose (the chlorite method, T9 m-54) were determined according to the standards of the Technical Association of Pulp and Paper Industry (TAPPI). The content and distribution of monosaccharides in agave fibres was determined by HPLC (Aminex HPX-87H; Bio Rad, Philadelphia, Pennsylvania, USA).

Acetosolv kinetics

Acetosolv experiments were carried out in roundbottom flasks with a solid/liquid ratio of 1/10. The aqueous solution with an 80% acetic acid concentration was heated to the boiling point, and when the target temperature was reached, a 0.3% hydrochloric acid concentration was added while stirring. The mixture was allowed to react for between 0 and 180 minutes. After the reaction, the pulp sample was filtered into medium-porosity glass crucibles and washed four times with acetic acid solutions. In the first of these washes, the quantity of the solution used was 2.5 times the dry weight of the agave fibres and, in the last two washes, the quantity of the solution was 5 times the dry weight. Finally, the pulp was repeatedly washed with distilled water until a neutral pH was acquired, and then the pulp was left to dry at room temperature. 33,34

Pulp characterisation

In order to study the kinetic changes, the pulp was characterised according to the Technical Association of Pulp and Paper Industry (TAPPI) standards. Moisture content was determined gravimetrically after drying the material at 105 °C (TAPPI T 412 om16). Pulp yield (PY) was determined gravimetrically and expressed as g dried pulp/100 g dry initial agave (% dry basis) after oven drying until constant weight was achieved. TAPPI standards were applied to determine the kappa number (KN) of pulp (T236 om99) and the viscosity index or pulp viscosity (VIS) in a cupriethylenediamine solution (T230 om99). The delignification selectivity (SEL) of pulping was calculated as:³⁵

$$Selectivity = \frac{Pulp \ viscosity}{Kappa \ number}$$
 (1)

Spectroscopic and microstructural characterisation

To investigate the influence of the Acetosolv treatment on the microstructure and composition of the agave fibres, a spectroscopic and microstructural

characterization of selected pulps was performed at different points during the process.

Confocal laser scanning microscopy (CLSM)

Confocal laser scanning microscopy permits to simultaneously monitor the changes on lignin and cellulose in the fibres during the Acetosolv process. Fibres or pulps were placed on a glass slide and observed under CLSM (LSM 710, Carl Zeiss, Germany). The presence of cellulose was examined during the pulping process by staining the fibres with calcofluor white M2R (fluorescent brightener 28 F3543, Sigma, USA) at 0.01% concentration for 5 minutes and then the fibres were dried at room temperature. Subsequently, the fibres were observed in the confocal microscope, using a diode laser (CW/pulsed) emitting at 405 nm. 36 Lignin was monitored through its autofluorescence and the fibres were observed at 488 nm using an Argon laser. 37,38 All images were acquired at 10x magnification in RGB color and stored in TIFF format at 512×512 pixels.

Scanning electron microscopy (SEM)

Microstructural changes occurring due to delignification were observed by SEM. Pulps were mounted in aluminum cylindrical sample holders provided with double sided carbon tape. The samples were observed with a scanning electron microscope (XL 30 ESEM, Philips, USA) at 300x and 25 kV.

Raman spectroscopy

Raman spectroscopy measurements were performed using a LabRam HR 800 (Horiba Jobin Yvon; Miyanohigashi, Kyoto, Japan) coupled to an Olympus BX 41 microscope with a 100x objective. Raman spectra were recorded using a 600 lines/mm grating and a 785 nm emission laser. Spectral resolution was approximately 2 cm⁻¹. The measurements were conducted using wavenumbers from 100 cm⁻¹ to 3060 cm⁻¹ (20 °C). The spectra were recorded using an exposure time of 8-10 s. The spectra acquisitions were managed by the LabSpec software (Horiba JobinYvon) and edited using the freeware Spekwin 32 program.³⁹

X-ray diffraction (XRD)

X-ray diffraction analysis was conducted using a Rikagu MiniFlex 600 diffractometer (Japan), with CuKα radiation source, at 40 kV and 15 mA. Scans were obtained from 3 to 60 degrees 2θ in 0.01 increments for 3 seconds per step. CrI was calculated as: $(I_{002}\text{-}I_{am})/(I_{002})$ after subtraction of the background signal measured without cellulose. A peak fitting program (PeakFit v.4.12) was used, assuming Lorentz functions for each peak and a broad peak at around 21.5° assigned to the amorphous contribution. Iterations were repeated until the maximum F number was obtained. In all the cases, the F number was >5000, corresponding to a R^2 value of 0.993.

RESULTS AND DISCUSSION

Chemical composition of agave fibres

The chemical composition of agave leaves is shown in Table 1. It can be seen that the ash content (3.5%) is within the range (2.1-8.0%)previously reported for fibre from raw cores. 46-48 Agave leaves contain 56-60% potentially fermentable sugars, over half of which are present in a soluble fraction. In the case of Agave atrovirens, the hot water extract content (43.5%) was higher than the one reported for untreated leaves of Agave americana (29.1%) and Agave tequilana (15.3%).⁴⁹ On the other hand, the acid insoluble lignin content was smaller than the one reported for untreated agave fibres $-2.1\%^{50}$ and 2.72%, ⁵¹ pineapple – 3.46%, ⁵² and leaf – 4.9%. ²² The holocellulose content (cellulose plus hemicelluloses) obtained in the present study (42.9%) is lower than the one reported for sisal leaves (49.6%) without treatment. 22,47,48 As for the distribution of monosaccharides in agave leaves, glucose was the majority sugar, whereas the noncellulosic sugar content (roughly hemicelluloses) reached a value of 11.9%. This content is similar to that reported for untreated leaves of A. americana (9.3%) and A. tequilana (9.7%). 49

Acetosoly kinetics

Figure 1 presents the response variables of the pulping process (PY, KN and VIS), which display an exponential behaviour. Figures 1A, 1B and 1C present the kinetics of delignification in the Acetosolv process under the aforementioned conditions with exponential decay behaviour. For KN and PY, a first stage can be seen during the first 30 minutes, corresponding to the rapid delignification phase that is explained by the easy removal of the surface lignin in the agave fibre; the second stage covers the lapse from the 30 min to 180 min; here, both KN and PY decreased at a lower speed compared to the first stage, which may be due to the extraction of interfibrillar lignin from the agave fibres (Fig. 1A and 1B). For the VIS, the first stage corresponds to the interval between 0 min and 40 min, while the second stage covers the lapse from 60 to 180 min. These stages of delignification were similar to those reported for the pulping of *Eucalyptus globulus*, ⁵³ and several hardwoods using a Kraft process. ⁵⁴ Table 2 presents the results of the pulping process. Selectivity is the ratio between viscosity and kappa number, thus a high delignification a high degree of selectivity indicates delignification with low cellulose degradation. 35,55

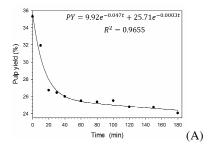
The results of Table 2 indicate better delignification selectivity of pulping at 30 (SEL = 95.5) and 40 (SEL = 102.2) min; however, if the optimal time of delignification is exceeded, selectivity decreases, which was confirmed by previous publications. 55,56 In general, organosolv pulping has a low effect on carbohydrate degradation (cellulose, in particular) and has good selectivity in the delignification reaction. The selectivity values of this work are higher than those reported for the Kraft process (69.2), the kraft process with anthraquinone (72.8), the kraft process with sodium xylenesulfonate pretreatment (83.4),the kraft process with xylenesulfonate pretreatment and addition of anthraquinone (85.3),35 and is well above the values for processes involving the use of microorganisms (15),⁵⁷ or processes involving ethanol and water (12-16).⁵⁸

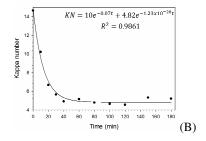
Microstructural changes due to pulping

In order to describe and evaluate the changes in agave fibres during the Acetosolv process, these fibres were analysed by means of SEM and CLSM (Fig. 2). The fluorescence of cellulose and lignin by CLSM is similar to that reported in previous studies, ^{6,28} in which it is assumed that the autofluorescence of timber materials is mainly due to the lignin content and its intensity is directly proportional to its concentration. SEM allows observing the agave fibre as a structure assembled in several layers, similar to the one reported by other authors for *A. americana* and *A. tequilana* Weber, blue variety. ^{36,46,59}

Table 1 Chemical composition of agave leaves

Component	Content (% dry basis)
Ash	3.5 ± 0.16
Hexane extracts	3.2 ± 0.13
Hot water extracts	43.5 ± 1.32
Acid insoluble lignin	1.9 ± 0.36
Holocellulose	42.9 ± 0.90
Monosaccharides (by HPLC)	
Glucose	35.7 ± 0.81
Xylose + galactose + mannose	10.4 ± 1.11
Arabinose	1.5 ± 0.55





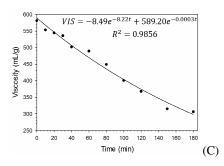


Figure 1: Acetosolv process kinetics: (A) pulp yield, (B) Kappa number and (C) viscosity index

Time	PY	ZN	VIS	CEI
(minutes)	(%)	KN	(mL/g)	SEL
0	35.3	14.6	581	39.8
10	31.9	10.2	553	54.2
20	26.7	6.7	544	81.2
30	26.4	5.6	535	95.5
40	26.0	4.9	501	102.2
60	25.5	5.1	489	95.9
80	25.4	4.8	449	93.5
100	25.5	4.6	400	87.0

4.6

5.3

5.2

24.8

24.7

24.1

Table 2
Response variables as a function of duration of Acetosolv process

At the beginning (0 min), the fluorescence of the cellulose was the lowest (Fig. 2F), and the autofluorescence of the lignin was very intense (CLSM), which corresponds to the lignified fibres with lower cellulose purification. Meanwhile, the SEM image illustrates the structural arrangement of the fibre bundle without damage (Fig. 2A).

120

150

180

The reaction continued until 30 minutes, when the fluorescence of the cellulose was the most intense (Fig. 3G), which indicates a purified fibre. Therefore, the autofluorescence of the lignin only appeared in small areas of the fibre in the CLSM image. In the SEM image (Fig. 2B), the defibrated fibre bundle can be observed; at 30 min, lignin dissolution occurred. This can be considered the optimal cellulose purification time, which correlates with the maximum value of CrI (Table 4). The time matches the inflection point of the two stages in KN and PY (Fig. 1A and 1B), which agrees with the results of other studies, where the rapid dissolution of materials occurred in the first minutes of the process. 2.60-62

As regards the 60 min duration of the reaction, the surface of the pulp is covered by lignin, indicating the condensation and reprecipitation of lignin on the surface (Fig. 2H). Also, in the SEM image (Fig. 2C), damage in the structure of the fibre can be observed.

After 100 minutes of reaction, the fibres demonstrated an increase in lignin autofluorescence (Fig. 2I). In the SEM image, compaction and damage of the fibres is observed (Fig. 2D). This behaviour is intensified after 180 minutes of reaction (Fig. 2E and 2J).

Raman spectroscopy

The pulps (treated fibres) of the Acetosolv process were analysed by FT-Raman

spectroscopy to provide a better understanding of the chemical changes occurring during the delignification of the fibres. Figure 3 presents the spectra of the five pulps considered (0, 30, 60, 100 and 180 min) and the untreated agave fibres after the baseline correction and the normalization of the area. Table 3 illustrates how some of the characteristics of composition can be distinguished, according to the assignments of Raman bands provided in previously published works. 63–66 It can be seen that strong bands lack intensity or have low intensity.

80.0

59.4

58.8

368

315

306

At 1,601 cm⁻¹, an intense band of lignin, which is attributed to the symmetric stretching of carbon-carbon double bonds, is absent in the spectra of 30 and 60 min pulps (Fig. 3), which indicates the absence of lignin in the fibres (Fig. 3). Another strong lignin band (1,660 cm⁻¹, C=O and C=C, which is attributed to the stretching of the coniferyl and sinapyl alcohols, as well as aldehyde-like structures) is absent in the spectra of all the pulps, whereas the lignin bands that are produced in the regions of 560-790 and 1,508-1,658 cm⁻¹ demonstrate slight intensities. Existing bands are due to vibrations in the carbohydrate structures. The bands at 1,096 and 1,122 cm⁻¹ (due to several stretches of glycosidic bonds) are present in the pulp (treated fibres) with greater intensity after 100 min. This could be interpreted as an increase in the delignification and condensation of lignin in the pulp after this period, with the subsequent deterioration of the cellulose (probably, hydrolytic) as the process conditions become more severe, which leads to the breaking of some glycosidic bonds with a concomitant decrease in the intensity of the bands generated in this part of the carbohydrate chain.

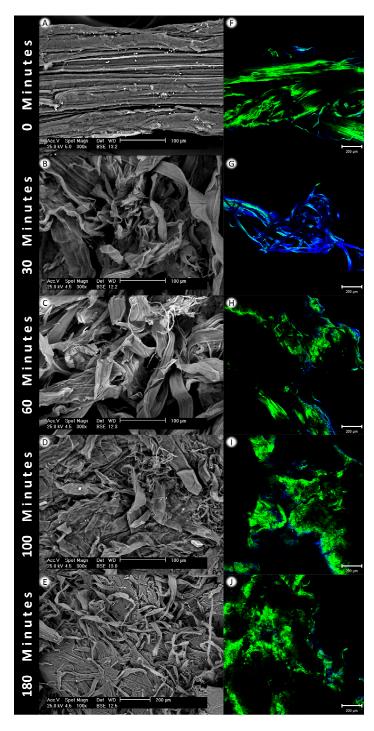


Figure 2: SEM images (A, B, C, D and E) and CLSM images (F, G, H, I and J) of agave fibre at different pulping time points (0, 30, 60, 100 and 1800 min); (green colour corresponds to lignin and blue colour to cellulose)

Table 3
Assignment of Raman bands in average spectra of agave fibres

Raman shift (cm ⁻¹)	Component	Assignment
2945	Lignin + Glucomannan	C-H asymmetric stretching in OCH ₃
2897	Cellulose	C-H stretching in R3C-H
1736	Hemicellulose	Carbonyl stretching vibrations of hemicelluloses
1658	Lignin	Stretching of conifer alcohol + stretching C=O in coniferaldehyde
1620	Lignin	C=C stretching of coniferaldehyde/sinapaldehyde
1601	Lignin	Aromatic ring symmetric stretching
1480	Cellulose	CH ₂ bending
1465	Cellulose	CH ₂ bending (scissors), COH bending in alcohols 1° and 2°
1423	Lignin	O-CH ₃ bending, ring stretching
1378	Cellulose	-CH ₃ bending, ring stretching
1330	Lignin	Aromatic ring breathing, CO stretching
1270	Lignin	Aryl-O in aryl-OH and aryl O-CH ₃ , vibration of guaiacyl ring with C=O grou
1130	Lignin	Vibration of conyferaldehyde/sinapaldehyde
1121	Cellulose + Xylan + Glucomannan	COC symmetric glycosidic stretching, ring breathing, skeletal vibration
1098	Cellulose + Xylan + Glucomannan	COC asymmetric glycosidic stretching, ring breathing, skeletal vibration
1037	Cellulose	CC, CO stretching. Alcohol 2°
896	Cellulose	HCC and HCO bending, cluster of peaks, primary methine bending
786	Lignin	Skeletal deformation
730	Lignin	Aromatic ring twisting
634	Lignin	Ring deformation
595	Lignin	Skeletal deformation
560	Lignin	Skeletal deformation
520	Cellulose	COC bending, glyc. links/CCC ring deformation
458	Cellulose	Bending CCO, CCC, ring deformation, skeletal bending
435	Cellulose	Bending CCO, CCC, ring deformation
378	Cellulose	Symmetric bending CC, ring deformation

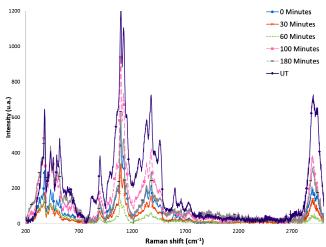


Figure 3: FT-Raman spectra of untreated agave fibres and pulps treated for different time periods

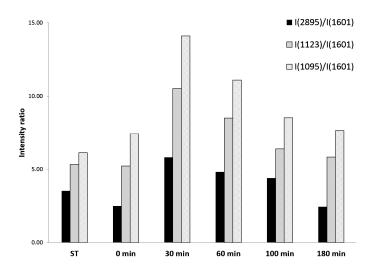


Figure 4: Comparison of intensity ratios of selected representative Raman bands of agave fibres subjected to Acetosolv pulping for different time periods

The characteristic cellulose bands (2,895, 1,123 and 1,095 cm⁻¹) reflect an enrichment of the solid cellulose. The previous discussion can be applied to almost all of these bands since they maintain the order of the intensities. Other bands are associated with symmetrical bending (380, 435, 460 cm⁻¹) and several vibrations of pitch (wagging) and bending (1,380 cm⁻¹) in the rings of glycopyranose.

Figure 4 presents a comparison between the ratios of the intensities of the characteristic bands of representative carbohydrates (1,095, 1,123 and 2,895 cm⁻¹) and the strongest band of lignin (1,601 cm⁻¹). The pulps reflect the characteristics of carbohydrates with the increase in the experimental variables, up to 30 minutes, because from this point, the degradation of the sugar

chains is predominant, which decreases all the intensity relationships.

X-ray diffraction analysis

Using the most common procedure for the measurement of the CrI,⁴⁰ namely, the peak height, the CrI was calculated and the data obtained are presented in Table 4. Figure 5 illustrates the X-ray diffractograms of the untreated agave fibres and those subjected to the Acetosolv pulping process for different time periods. The effect of pulping duration on the pulps was evidenced after 60 minutes of reaction, when the crystallinity is diminished in the X-ray diffractograms, which is evident in the difference in the relative intensities of the peaks analysed. The CrI calculated from the X-ray data

determining these peaks is given in Table 4, where CrI increases in the first 30 minutes of the treatment; the highest crystallinity value was obtained for the 30 min pulping. XRD analysis reveals that treated agave fibres seem to have undergone some polymorphic changes, upon the Acetosolv treatment, on the principal diffraction planes, $101 (2\theta = 15.4-17.7^{\circ})$, $002 (2\theta = 22.0-19.100)$

 23.4°) and $040~(2\theta = 34.1-35.9^{\circ})$, which gradually turned to minimum values over treatment time. However, the diffractograms do not demonstrate a clear division of the bands (especially, plane 002), which implies that cellulose I (native) was converted to cellulose II.

Table 4
Crystalline index of treated agave leaves determined by Segal's method

Dula	Crystalline index (CrI, %)		
Pulp	Peak height		
UT	43.2		
0 minutes	61.9		
30 minutes	70.3		
60 minutes	69.5		
100 minutes	68.6		
180 minutes	66.2		

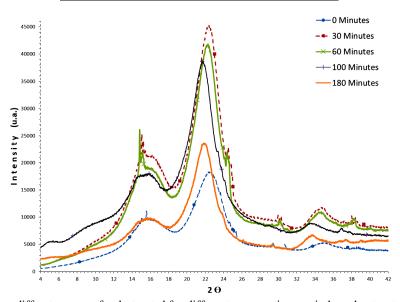


Figure 5: X-ray diffractograms of pulp treated for different process time periods and untreated agave pulp

CONCLUSION

This work describes the changes occurring in agave fibres during a green pulping process and evaluates them by microscopy and spectroscopy techniques. providing a comprehensive quantitative view of the phenomenon occurring during the delignification of agave fibres. The delignification process occurs in two stages: the first one corresponds to the superficial delignification of the fibre (0-30 min), and the second stage corresponds to the removal interfibrillar lignin of the fibre, which is corroborated by the disappearance of the

characteristic band of lignin in Raman spectra and the increase in the index of crystallinity. The microscopy techniques applied in this work were useful to monitor the microstructural changes occurring during the pulping processing, while FT-Raman spectroscopy confirmed the data obtained by chemical analysis about pulp purity and the efficiency of the delignification process in terms of cellulose preservation in the fibre.

REFERENCES

¹ H.-L. Li, D. She, P. Peng, Q. Xu, J. Liu *et al.*, *Cellulose Chem. Technol.*, **51**, 433 (2017).

- ² A. de Vega, P. Ligero, A. De Vega and P. Ligero, *Ind. Crop. Prod.*, **97**, 252 (2017).
- ³ N. F. Souza, J. A. Pinheiro, A. I. S. Brígida, J. P. S. Morais, M. M. de Souza Filho *et al.*, *Ind. Crop. Prod.*, **94**, 480 (2016).
- ⁴ T. Virtanen, K. Svedström, S. Andersson, L. Tervala, M. Torkkeli *et al.*, *Cellulose*, **19**, 219 (2012).
- ⁵ R. Moriana, F. Vilaplana and M. Ek, *Cellulose*, **22**, 3409 (2015).
- ⁶ H. M. Hernández-Hernández, J. J. Chanona-Pérez, A. Vega, P. Ligero, J. A. Mendoza-Pérez *et al.*, *Ind. Crop. Prod.*, **86**, 163 (2016).
- ⁷ M. George, W. Shen, K. Sharma, C. Montemagno and M. George, *Cellulose Chem. Technol.*, **51**, 681 (2017).
- ⁸ N. Marin, A. C. Puitel, A.-M. Chesca and D. Gavrilescu, *Cellulose Chem. Technol.*, **51**, 745 (2017).
- ⁹ R. C. Saxena, D. K. Adhikari and H. B. Goyal, *Renew. Sustain. Energ. Rev.*, **13**, 167 (2009).
- ¹⁰ T. Joffre, O. Girlanda, F. Forsberg, F. Sahlén, M. Sjödahl *et al.*, *Cellulose*, **22**, 2993 (2015).
- M. Jonoobi, R. Oladi, Y. Davoudpour, K. Oksman, A. Dufresne *et al.*, *Cellulose*, **22**, 935 (2015).
- ¹² A. L. de Lemos, C. J. Mauss and R. M. C. Santana, *Cellulose Chem. Technol.*, **51**, 711 (2017).
- A. Z. M. Salem, N. T. Valdez, O. A. Olafadehan, M. M. Y. Elghandour, A. Barbabosa Pliego *et al.*, *J. Equine Vet. Sci.*, 48, 103 (2017).
- ¹⁴ C. M. Y. Huang, P. X. Chia, C. S. S. Lim, J. Q. Nai, D. Y. Ding *et al.*, *Cellulose Chem. Technol.*, **51**, 665 (2017).
- ¹⁵ B. V. M. Rodrigues, E. C. Ramires, R. P. O. Santos and E. Frollini, *J. Appl. Polym. Sci.*, **132**, 16 (2015).
- ¹⁶ A. Mandal and D. Chakrabarty, *J. Ind. Eng. Chem.*, **20**, 462 (2014).
- ¹⁷ C. J. Huntley, K. D. Crews and M. L. Curry, *Int. J. Polym. Sci.*, **2015**, 1 (2015).
- B. Deepa, E. Abraham, N. Cordeiro, M. Mozetic,
 A. P. Mathew *et al.*, *Cellulose*, 22, 1075 (2015).
- ¹⁹ D. M. do Nascimento, J. S. Almeida, M. do S. Vale, R. C. Leitão, C. R. Muniz *et al.*, *Ind. Crop. Prod.*, **93**, 66 (2016).
- ²⁰ J. R. González-López, J. F. Ramos-Lara, A. Zaldivar-Cadena, L. Chávez-Guerrero, R. X. Magallanes-Rivera *et al.*, *Fuel Process. Technol.*, **133**, 35 (2015).
- ²¹ J. R. Mielenz, M. Rodriguez, O. A. Thompson, X. Yang and H. Yin, *Biotechnol. Biofuels*, **8**, 79 (2015).
- ²² S. C. Davis and S. P. Long, in "Industrial Crops: Breeding for BioEnergy and Bioproducts", New York, Springer, 2015, pp. 335-349.
- ²³ G. Iñiguez-Covarrubias, R. Díaz-Teres, R. Sanjuan-Dueñas, J. Anzaldo-Hernández and R. M. Rowell, *Bioresour. Technol.*, **77**, 101 (2001).
- ²⁴ A. Ferrer, A. Vega, A. Rodríguez and L. Jiménez, *Bioresour. Technol.*, **132**, 115 (2013).
- ²⁵ Q. Tu, S. Fu, H. Zhan, X. Chai and L. A. Lucia, *J. Agric. Food Chem.*, **56**, 3097 (2008).
- ²⁶ V. P. Soudham, D. Rodriguez, G. J. M. Rocha, M.

- J. Taherzadeh and C. Martin, For. Stud. China, 13, 64 (2011).
- ²⁷ M. Zhang, K. S. Hui, G. Gong, K. N. Hui and L. Liu, *Cellulose Chem. Technol.*, **51**, 425 (2017).
- ²⁸ W. Wanrosli, I. Mazlan, K. Law and R. Nasrullah, *Maderas. Cienc. Tecnol.*, **13**, 193 (2011).
- ²⁹ J. F. Ma, G. H. Yang, J. Z. Mao and F. Xu, *Ind. Crop. Prod.*, **33**, 358 (2011).
- ³⁰ Q. Yang and X. Pan, *Bioresour. Technol.*, **126**, 336 (2012).
- ³¹ M. F. Li, S. Yang and R. C. Sun, *Bioresour*. *Technol.*, **200**, 971 (2016).
- ³² H. M. Hernández-Hernández, J. J. Chanona-Pérez, A. Vega, P. Ligero, R. R. Farrera-Rebollo *et al.*, *Microsc. Microanal.*, 22, 5 (2016).
- ³³ C. Gumeta-Chávez, J. J. Chanona-Pérez, J. A. Mendoza-Pérez, E. Terrés-Rojas, V. Garibay-Febles *et al.*, *Dry. Technol.*, **29**, 612 (2011).
- ³⁴ P. Ligero, J. J. Villaverde, A. Vega and M. Bao, *Ind. Crop. Prod.*, **25**, 294 (2007).
- ³⁵ P. Ligero, A. Vega and M. Bao, *Ind. Crop. Prod.*, 21, 235 (2005).
- ³⁶ A. Bessadok, D. Langevin, F. Gouanvé, C. Chappey, S. Roudesli *et al.*, *Carbohyd. Polym.*, **76**, 74 (2009).
- ³⁷ F. Xu, X. C. Zhong, R. C. Sun and Q. Lu, *Ind. Crop. Prod.*, **24**, 186 (2006).
- ³⁸ K. H. Lee, S. G. Wi, A. P. Singh and Y. S. Kim, *J. Wood Sci.*, **50**, 281 (2004).
- ³⁹ Z. Zhang, J. Ma, Z. Ji and F. Xu, *Microsc. Microanal.*, **18**, 1459 (2012).
- ⁴⁰ Y. Hu, L. Tang, Q. Lu, S. Wang, X. Chen and B. Huang, *Cellulose*, **21**, 1611 (2014).
- D. Yang, L. X. Zhong, T. Q. Yuan, X. W. Peng and R. C. Sun, *Ind. Crop. Prod.*, **43**, 141 (2013).
- ⁴² S. Park, J. O. Baker, M. E. Himmel, P. A. Parilla and D. K. Johnson, *Biotechnol. Biofuels*, **3**, 10 (2010).
- ⁴³ C. Driemeier, M. T. B. Pimenta, G. J. M. Rocha, M. M. Oliveira, D. B. Mello *et al.*, *Cellulose*, **18**, 1509 (2011).
- ⁴⁴ L. Segal, J. J. Creely, J. A. E. Martin and C. M. M. Conrad, *Text. Res. J.*, **29**, 786 (1959).
- ⁴⁵ E. L. Hult, T. Iversen and J. Sugiyama, *Cellulose*, **10**, 103 (2003).
- ⁴⁶ C. J. Garvey, I. H. Parker and G. P. Simon, *Macromol. Chem. Phys.*, **206**, 1568 (2005).
- ⁴⁷ J. He, S. Cui and S. Wang, J. Appl. Polym. Sci., 107, 1029 (2008).
- ⁴⁸ S. Kestur, T. H. S. Flores-Sahagun, L. P. Dos Santos, J. Dos Santos, I. Mazzaro *et al.*, *Appl. Sci. Manuf.*, **45**, 153 (2013).
- ⁴⁹ C. S. S. S. Lima, M. M. Conceição, F. L. H. H. Silva, E. E. Lima, L. S. Conrado *et al.*, *Appl. Energ.*, **102**, 254 (2013).
- ⁵⁰ L. Yang, M. Lu, S. Carl, J. A. Mayer, J. C. Cushman *et al.*, *Biomass. Bioenerg.*, **76**, 43 (2015).
- ⁵¹ K. R. Corbin, C. S. Byrt, S. Bauer, S. DeBolt, D. Chambers *et al.*, *PLoS One*, **10**, 1 (2015).
- ⁵² A. Altinisik, Y. Seki, S. Ertas, E. Akar, E. Bozacı

- et al., Fiber. Polym., **16**, 370 (2015).

 A. E. O. Ben Sghaier, Y. Chaabouni, S. Msahli and F. Sakli, *Ind. Crop. Prod.*, **36**, 257 (2012).
- ⁵⁴ B. M. Cherian, A. L. Leão, S. F. de Souza, S. Thomas, L. A. Pothan et al., Carbohyd. Polym., 81, 720 (2010).
- ⁵⁵ A. Santos, F. Rodriguez, M. Gilarranz, D. Moreno and F. Garcia-Ochoa, Ind. Eng. Chem. Res., 36, 4114
- (1997). ⁵⁶ R. B. Santos, E. A. Capanema, M. Y. Balakshin, H. M. Chang and H. Jameel, J. Agric. Food Chem., 60,
- 4923 (2012). ⁵⁷ A. S. Singha and R. K. Rana, *Carbohyd. Polym.*, **87**, 500 (2012).
- ⁵⁸ J. J. Villaverde, J. Li, M. Ek, P. Ligero and A. De Vega, J. Agric. Food Chem., 57, 6262 (2009).

- H. M. Hernández-Hernández, J. J. Chanona-Pérez, G. Calderón-Domínguez, M. J. Perea-Flores, J. A. Mendoza-Pérez et al., Microsc. Microanal., 20, 1436 (2014).
- V. C. Coletta, C. A. Rezende, F. R. da Conceição, I. Polikarpov and F. E. G. Guimarães, Biotechnol. Biofuels, 6, 43 (2013).
- K. Kavkler and A. Demšar, Spectrochim. Acta A, **78**, 740 (2011).
- U. P. Agarwal and S. A. Ralph, Appl. Spectrosc., **51**, 1648 (1997).
- U. P. Agarwal and R. S. Reiner, J. Raman Spectrosc., **40**, 1527 (2009).
- U. P. Agarwal, J. D. McSweeny and S. A. Ralph, J. Wood Chem. Technol., 31, 324 (2011).